# Noise in Gene Regulatory Networks 

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## Summary

Models for Chemical Reactions<br>Macroscopic Model<br>Microscopic Model<br>Mesoscopic Model

Chemical Reactions as Markov Processes
The Chemical Master Equation
A result for linear/affine transitions
An equivalence property

Comments

## Stochastic and Deterministic Models

Chemical reactions can be modeled by

- Deterministic ordinary differential equations involving the species concentrations (macroscopic model)
- Many spatial agents each one describing a single molecule (microscopic model)

Both the approaches present advantages and drawbacks

- When there are few molecules of a species the concentration description is meaningless, mass action kinetics are not valid, and discrete models are necessary.
- Multi-agent models require a huge computational effort to be simulated


## Macroscopic model

The system state is given by the concentration of the species and ODEs can be easily written.

Consider the reversible reactions

$$
\begin{align*}
& S_{1} \xrightarrow{K_{21}} S_{2}  \tag{1}\\
& S_{2} \xrightarrow{K_{12}} S_{1} \tag{2}
\end{align*}
$$

leading to the ODEs

$$
\begin{array}{r}
\dot{x}_{1}=-K_{21} x_{1}+K_{12} x_{2} \\
\dot{x}_{2}=K_{21} x_{1}-K_{12} x_{2} \tag{4}
\end{array}
$$

where $x_{i}$ is the concentration of the species $S_{i}$.

## Microscopic model

Every single particle is modeled as a distinct agent with a specific position and velocity.


- Monomolecular reactions occur according to the given rate.
- Dimolecular reactions occur when two particles are in the same spot and the absolute value of their relative velocity is large enough.


## The Chemical Master Equation (Mesoscopic model)

The mesoscopic model tries to combine both the approaches.

- The system is described in terms of the $n$ reactants and/or products $\left\{S_{1}, S_{2}, \ldots, S_{n}\right\}$
- The state of the system is given by the integer vector $\left\{x_{1}, x_{2}, \ldots, x_{n}\right\}$ representing the number of molecules for every species
- The system is considered a time-continuous Markov model
- The $M$ chemical reactions represent the Markov transitions which can be described by the stoichometric vectors $\left(r_{1}, \ldots, r_{M}\right)$
- The probabilities of every state configuration are described by a vector $\mathcal{P}(t)=\left[\mathcal{P}_{1}(t), \mathcal{P}_{2}(t), \ldots\right]$ which evolves according to

$$
\begin{equation*}
\dot{\mathcal{P}}(t)=A \mathcal{P}(t) \tag{5}
\end{equation*}
$$

A graphical representation clarifies the meaning of the Mesoscopic Model


The reactions with the relative stoichometric vectors are

$$
\begin{array}{lr}
\emptyset \rightarrow S_{1} & r_{1}=(1,0) \\
S_{1}+S_{2} \rightarrow S_{1} & r_{2}=(0,-1) \\
S_{1} \rightarrow \emptyset & r_{3}=(-1,0)
\end{array}
$$

## Transition Probabilities

The probability that the reaction $R_{\mu}$ will be the next reaction and that will occur within the next time $d t$ is given by $a_{\mu}(x) d t$.

- If $R_{\mu}$ represents a supply $\left(\emptyset \rightarrow{ }^{\prime}\right.$ Products'), then

$$
\begin{equation*}
a_{\mu}(x)=c_{\mu} \tag{6}
\end{equation*}
$$

- If $R_{\mu}$ is a monomolecular reaction $\left(S_{i} \rightarrow{ }^{\prime}\right.$ Products'), then

$$
\begin{equation*}
a_{\mu}(x)=c_{\mu} x_{i} \tag{7}
\end{equation*}
$$

- If $R_{\mu}$ is a dimolecular reaction $\left(S_{i}+S_{j} \rightarrow{ }^{\prime}\right.$ Products'), then

$$
\begin{equation*}
a_{\mu}(x)=c_{\mu} x_{i} x_{j} \tag{8}
\end{equation*}
$$

## Chemical Reactions and Markov Processes

Let us consider a chemical system with $n$ molecular species and $m$ elementary reactions and its related mesoscopic model.

$$
\begin{equation*}
x(t) \xrightarrow{W_{i}(x(t))} x(t)+r_{i} \tag{9}
\end{equation*}
$$

where $x(t) \in \mathbb{N}^{n}$ is the vector of molecule numbers at time $t$; $r_{i} \in \mathbb{N}^{n}$ is the stechiometric vector of the $i$-th reaction and $W_{i} \in \Re^{+}$is the transition rate.
The transitions described by (9) can be naturally modeled by a jump Markov system (markovian Petri net).

## The Chemical Master Equation

Let us consider how the probability $p$ of the configuration $x$ at time $t$ evolves. It satisfies the ODE

$$
\begin{equation*}
\dot{p}(t)(x, t)=\sum_{i} W_{i}\left(x-r_{i}\right) p\left(x-r_{i}, t\right)-\sum_{i} W_{i}(x) p(x, t) \tag{10}
\end{equation*}
$$

which is known as the Chemical Master Equation.
Defining $\mathcal{P}(t)$ as the vector containing the probabilities of all the possible configurations $x$, we have that

$$
\begin{equation*}
\dot{\mathcal{P}}(t)=\mathcal{A} \mathcal{P} \tag{11}
\end{equation*}
$$

where the entries in $\mathcal{A}$ depend on the terms $W_{i}(x)$. This means that the evolution of the PDF follows a linear dynamics, even though the transition rates are not linear.

## Notation

- I shall use capital letters to denote random variables/random processes and lower case letter to denote their realizations
- $E[X]$ is the mean of the stochastic variable $X$
- $E_{X}(t):=E[X(t)]$ is the mean of the stochastic process $X(t)$ at time $t$
- $V_{X Y}\left(t_{1}, t_{2}\right):=E\left[X\left(t_{1}\right) X\left(t_{2}\right)^{T}\right]$ is the second moment
- $R_{X Y}\left(t_{1}, t_{2}\right):=E\left[X\left(t_{1}\right) X\left(t_{2}\right)^{T}\right]-E\left[X\left(t_{1}\right)\right] E\left[X\left(t_{2}\right)\right]$ is the second central moment
- $R_{X}\left(t_{1}, t_{2}\right):=R_{X X}\left(t_{1}, t_{2}\right)$ and $V_{X}\left(t_{1}, t_{2}\right):=V_{X X}\left(t_{1}, t_{2}\right)$


## Notation

A process $X$ is stationary (wide sense) if and only if, for all $t, t_{1}, t_{2}, \tau$,

$$
\begin{equation*}
E_{X}(t)=E_{X} ; \quad R_{X}\left(t_{1}, t_{2}\right)=R_{X}\left(t_{1}+\tau, t_{2}+\tau\right) \tag{12}
\end{equation*}
$$

Two stationary processes $X$ and $Y$ are mutually stationary (wide sense) if and only if

$$
\begin{equation*}
R_{X Y}\left(t_{1}, t_{2}\right)=R_{X Y}\left(t_{1}+\tau, t_{2}+\tau\right) \tag{13}
\end{equation*}
$$

In such a context, the following definitions make sense

- $R_{X Y}(t):=R_{X Y}(t, 0)$ (abusing the notation!)
- $\Phi_{X Y}(i \omega):=\mathcal{F}\left(R_{X Y}(t, 0)\right)(i \omega)$


## A result for linear/affine transitions

In many applicative cases, the transition rate functions $W_{i}(\cdot)$ can be considered linear. In the case of linear/affine transition rates some analytical results can be proved.
Theorem
If $W_{i}(x)=c_{1}^{(i)} x+c_{0}^{(i)}$ then, there exist matrices
$A, D(t)=D\left(E_{X}(t)\right) \in \Re^{n \times n}, D(t)$ semi-positive defined and a vector $C_{0} \in \Re^{n}$ such that

$$
\begin{align*}
& \frac{d}{d t} E_{X}(t)=A E_{X}(t)+C_{0}  \tag{14}\\
& \frac{d}{d t} R_{X}(t, t)=A \frac{d}{d t} R_{X}(t, t)+\frac{d}{d t} R_{X}(t, t)^{T} A^{T}+D\left(E_{X}(t)\right)  \tag{15}\\
& \frac{d}{d t_{1}} R_{X}\left(t_{1}, t_{2}\right)=A R_{X}\left(t_{1}, t_{2}\right) \tag{16}
\end{align*}
$$

## Proof

Let us evaluate $\frac{d}{d t} E_{X}(t)$

$$
\begin{aligned}
\frac{d}{d t} E_{X}(t) & =\sum_{x} x \frac{d p(x, t)}{d t}= \\
& =\sum_{i, x} x W_{i}\left(x-r_{i}\right) p\left(x-r_{i}, t\right)-\sum_{i, x} x W_{i}(x) p(x, t)= \\
& =\sum_{i, x}\left(x+r_{i}\right) W_{i}(x) p(x, t)-\sum_{i, x} x W_{i}(x) p(x, t)= \\
& =\sum_{i, x} r_{i} W_{i}(x) p(x, t)
\end{aligned}
$$

Define $f(x):=\sum_{i} r_{i} W_{i}(x)$ and we have

$$
\frac{d}{d t} E_{X}(t)=E[f(x(t))]
$$

## Proof

In the linear/affine case $W_{i}(x)=c_{1}^{(i)} x+c_{0}^{(i)}$, thus

$$
\begin{align*}
f(x(t))= & \sum_{i} r_{i} W_{i}(x(t))=\sum_{i} r_{i}\left[c_{1}^{(i)} x(t)+c_{0}^{(i)}\right]=  \tag{17}\\
& =\sum_{i} r_{i} c_{1}^{(i)} x+\sum_{i} r_{i} c_{0}^{(i)}=A x(t)+C_{0} \tag{18}
\end{align*}
$$

That leads to

$$
\begin{aligned}
\frac{d}{d t} E_{X}(t) & =E[f(x(t))]=E\left[A x(t)+C_{0}\right] \\
& =A E_{X}(t)+C_{0}
\end{aligned}
$$

## Proof

Let us evaluate the dynamics of $R_{X X}(t, t)$

$$
\begin{aligned}
& \frac{d}{d t} V_{X}(t, t)=\sum_{x} x x^{T} \frac{d p(x, t)}{d t}= \\
& \quad=\sum_{x, i} x x^{T}\left[W_{i}\left(x-r_{i}\right) p\left(x-r_{i}, t\right)-W_{i}(x) p(x, t)\right]= \\
& \\
& =\sum_{x, i}\left(x+r_{i}\right)\left(x+r_{i}\right)^{T} W_{i}(x) p(x, t)-\sum_{x, i} x x^{T} W_{i}(x) p(x, t)= \\
& \\
& =E\left[\sum_{i} x r_{i}^{T} W_{i}(x)\right]+E\left[\sum_{i} r_{i} x^{T} W_{i}(x)\right]+E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x)\right]= \\
& \\
& \quad=E\left[x f^{T}(x)\right]+E\left[f(x) x^{T}\right]+E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x)\right]
\end{aligned}
$$

## Proof

$$
\begin{aligned}
& \frac{d}{d t} R_{X}(t, t)=\frac{d}{d t} V_{X}(t, t)-\frac{d}{d t}\left(E_{X}(t) E_{X}(t)^{T}\right)= \\
& \quad=E\left[x(t) f^{T}(x(t))\right]+E\left[f(x(t)) x(t)^{T}\right]+E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x(t))\right]+ \\
& \quad-E_{X}(t) E\left[f(x(t))^{T}\right]-E[f(x)] E_{X}(t)^{T}= \\
& =E\left[\left(x(t)-E_{X}(t)\right)(f(x(t))-E[f(x(t))])^{T}\right]+ \\
& \quad+E\left[(f(x(t))-E[f(x(t))])\left(x(t)-E_{X}(t)\right)^{T}\right]+ \\
& \quad+E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x(t))\right]
\end{aligned}
$$

## Proof

In the linear case $f(x(t))-E[f(x(t))]=A\left(x(t)-E_{X}(t)\right)$, thus

$$
\begin{equation*}
\frac{d}{d t} R_{X}(t, t)=A R_{X}(t)+R_{X}(t) A+E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x(t))\right] \tag{19}
\end{equation*}
$$

Define $D(t)$ and consider its Choleski factorization in terms of $B(t)$, as follows

$$
\begin{align*}
& D(t)=D(E[W(x(t))]):=E\left[\sum_{i} r_{i} r_{i}^{T} W_{i}(x(t))\right]  \tag{20}\\
& D(t)=B(t) B(t)^{T} \tag{21}
\end{align*}
$$

We can write that

$$
\begin{equation*}
\frac{d}{d t} R_{X}(t, t)=A R_{X}(t)+R_{X}(t) A+D(t) \tag{22}
\end{equation*}
$$

## Proof

Let $p\left(x_{1}, t_{1}, x_{2}, t_{2}\right)$ be the joint probability of the random variables $X\left(t_{1}\right)$ and $X\left(t_{2}\right)$. Let us evaluate the dynamics of $R_{X X}\left(t_{1}, t_{2}\right)$, observing that the Chemical Master Equation holds also for the joint probability.

$$
\begin{aligned}
\frac{d}{d t_{1}} & V_{X}\left(t_{1}, t_{2}\right)=\sum_{i, x, y} x y^{T} \frac{d}{d t_{1}} p\left(x_{1}, t_{1}, x_{2}, t_{2}\right)= \\
& =\sum_{x, y} x y^{T}\left[W_{i}\left(x-r_{i}\right) p\left(x-r_{i}, t_{1}, y, t_{2}\right)-W_{i}(x) p\left(x, t_{1}, y, t_{2}\right)\right] \\
& =\sum_{i, x, y}\left(x+r_{i}\right) y^{T} W_{i}(x) p\left(x, t_{1}, y, t_{2}\right)-\sum_{i, x, y} x y^{T} W_{i}(x) p\left(x, t_{1}, y, t_{2}\right) \\
& =\sum_{i, x, y} r_{i} W_{i}(x) p\left(x, t_{1}, y, t_{2}\right)
\end{aligned}
$$

## Comments

In the non-stationary case we have

$$
\begin{align*}
& \frac{d}{d t} E_{X}(t)=A E_{X}(t)+C_{0}  \tag{23}\\
& \frac{d}{d t} R_{X}(t, t)=A R_{X}(t, t)+R_{X}(t, t) A^{T}+D\left(E_{X}(t)\right)  \tag{24}\\
& \frac{d}{d t_{1}} R_{X}\left(t_{1}, t_{2}\right)=A R_{X}\left(t_{1}, t_{2}\right) \tag{25}
\end{align*}
$$

The evolution of the mean of the molecule numbers follows a standard linear/affine dynamics.
The autocovariance of the molecule numbers evolves according to the second equation which is a Differential Riccati Equation. The cross-covariance follows a purely linear dynamics.

## The stationary scenario

We have that the signal $E_{X}$ does not depend on $t$, therefore it satisfies the relation

$$
\begin{equation*}
A E_{X}+C_{0}=0 \tag{26}
\end{equation*}
$$

This implies that $D(t)=D\left(E_{X}(t)\right)$ does not depend on $t$, either. Moreover, $R_{X}\left(t_{2}+\tau, t_{2}\right)$ depends upon $\tau$ only. Hence we have

$$
\begin{align*}
& A R_{X}(t, t)+R_{X}(t, t) A^{T}+D=0  \tag{27}\\
& \frac{d}{d \tau} R_{X}(t+\tau, t)=A R_{X}(t+\tau, t) \tag{28}
\end{align*}
$$

That Lyapunov Equation is known in physics as the fluctuation-dissipation problem, since it is very often related to a balance between phenomena of expansion and dissipation.

## An equivalence property

Theorem
Assume that the Markov process described by the reactions $x(t) \xrightarrow{W_{i}(x(t))} x(t)+r_{i}$ with affine transition rates reaches the steady state. Consider

$$
\begin{align*}
& A R_{X}(t, t)+R_{X}(t, t) A^{T}+D=0  \tag{29}\\
& \frac{d}{d \tau} R_{X}(t+\tau, t)=A R_{X}(t+\tau, t) \tag{30}
\end{align*}
$$

Factorize $D$ as $D=B B^{T}$ The power spectrum of $x$ is given by

$$
\begin{equation*}
\Phi_{X}(\omega)=G(i \omega) G(i \omega)^{*} \tag{31}
\end{equation*}
$$

where $G(s)=(s I-A)^{-1} B$

